#### **Pool-boiling Experiments**

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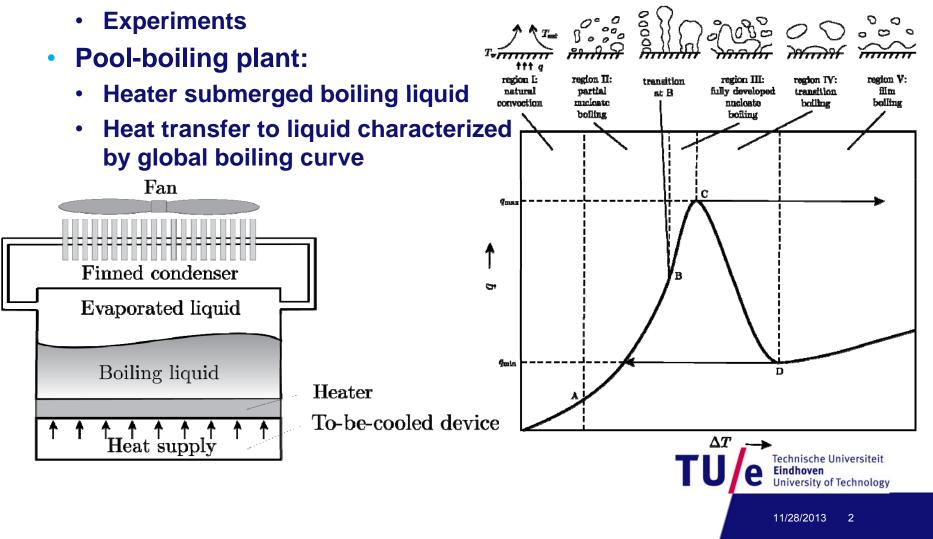
# Index

- Introduction
- Objective
- Background
- Experimental setup
- Experiments
- Conclusions



### Introduction

Can theoretical phenomena be observed in a practical setup?



# Objective

Can theoretical phenomena be observed in a practical setup?

- Field of application: EVs
- Start small scale: submerge 1 battery in boiling liquid

#### **Objective:**

 Experimental investigation of the ability of pool boiling for thermal conditioning of batteries in EVs

First exploratory experiments are carried out to investigate:

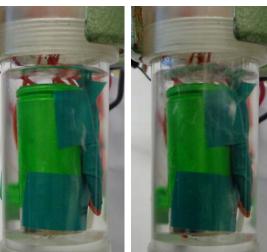
- The electric interaction between battery and cooling liquid
- The cooling capacity of the proposed cooling liquid
- The ability of the boiling process to thermally homogenize batteries
- The controllability of the boiling process

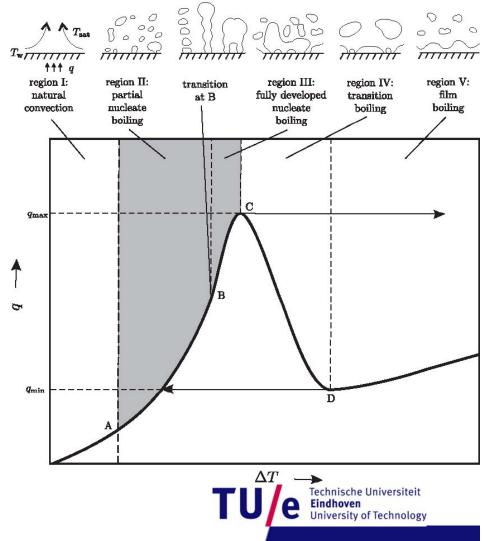


# Background

- Due to limited heat generation by battery only small region of boiling curve will be touched
  - Region I: Natural convection
  - Region II: Partial nucleate boiling
  - Region III: fully developed nucleate boiling







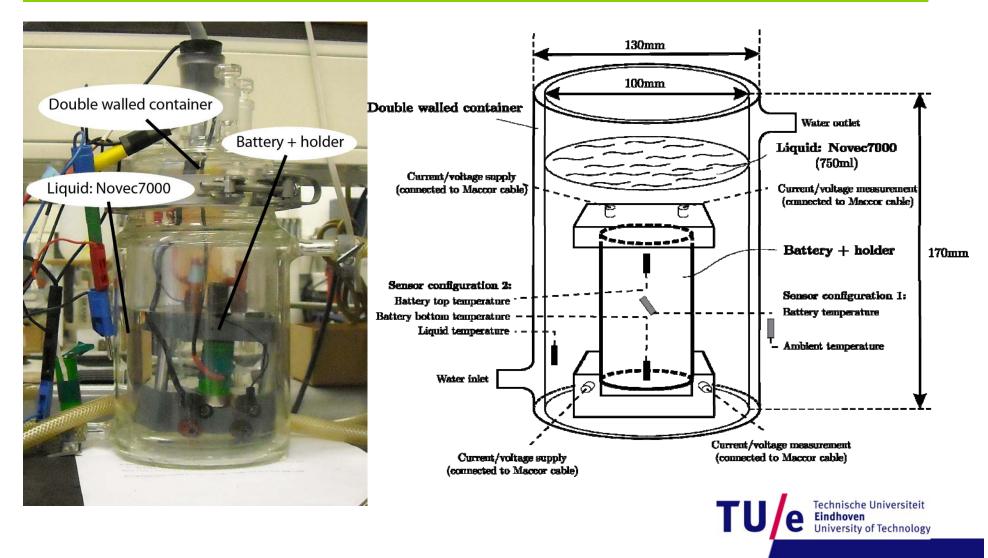
# **Experimental setups**

Two setups are utilized for the experiments

- In cooperation with dept. of Chemical Engineering and Chemistry
- Battery testing system (Maccor 2300, USA)
- Battery: Sony US18500VR (d = 18mm, h = 49mm)
  - 1Ah Li-ion battery
- Working fluid: Novec7000
  - 3M, USA, chemical composition: 99.5 weight percentage of C<sub>3</sub>F<sub>7</sub>OCH<sub>3</sub> (1methoxyheptafluoropropane)
  - Boiling point @ atm pressure = 34 °C
- The first setup (setup 1) is used to investigate:
  - the dielectric properties of the fluid.
  - discharge experiments at atmospheric pressure.
  - pulse-charge-discharge (PCD) experiments at atmospheric pressure
- The second setup (setup 2) is used to investigate:
  - pressure variations on the boiling process



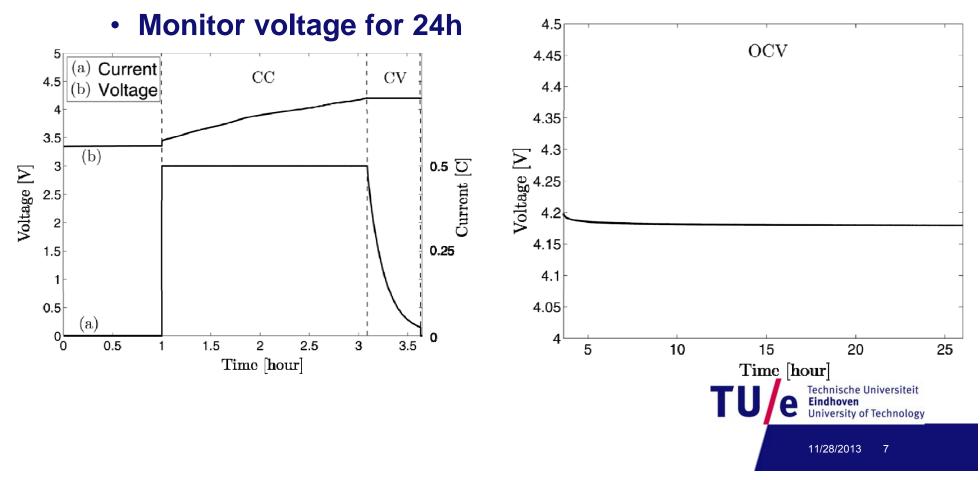
# Setup 1



11/28/2013 6

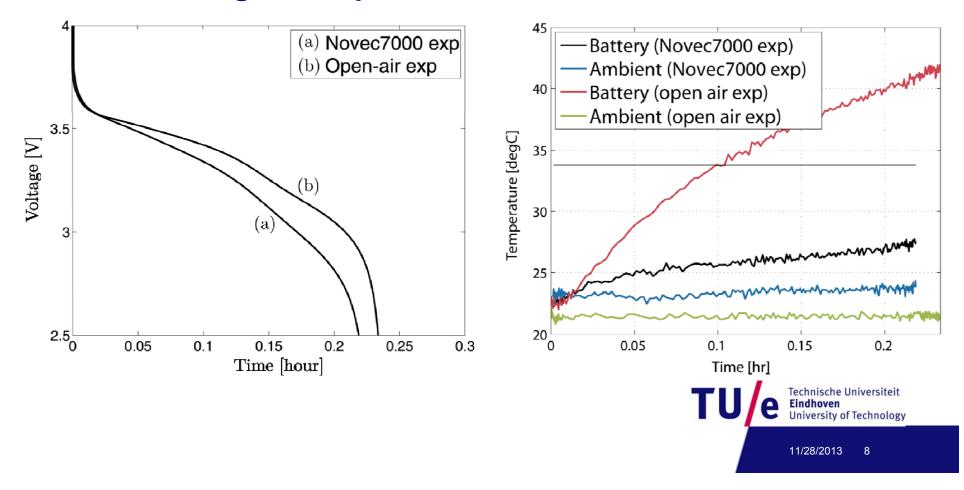
#### **Experiments:** Dielectric property of the working fluid

- Submerge battery in liquid
  - Fully load battery

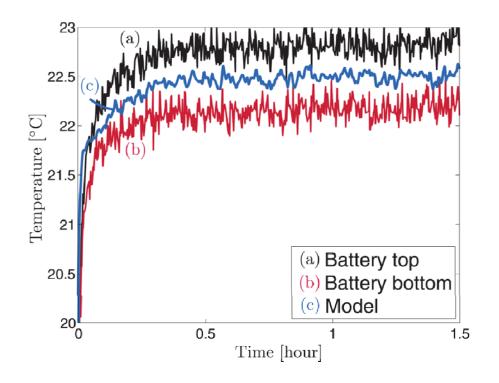


### Experiments: Submerged discharging

- **Discharge** @ 5A (maximal current by Maccor system)
  - submerged in liquid vs in open-air



- Non boiling regime
  - $T_{wall} = 20 \, {}^{\circ}C$

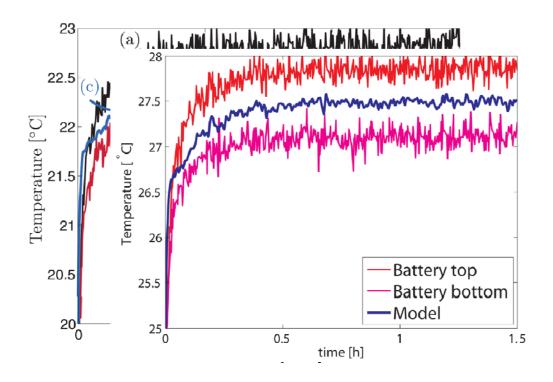


Boiling regime



**Boiling regime** 

- Non boiling regime
  - T<sub>wall</sub> = 25 °C

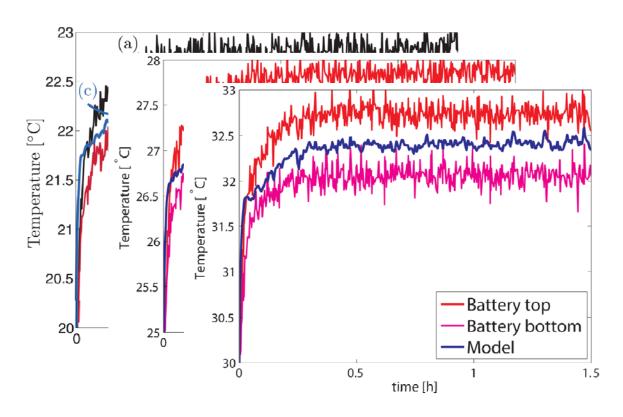




Non boiling regime

Boiling regime

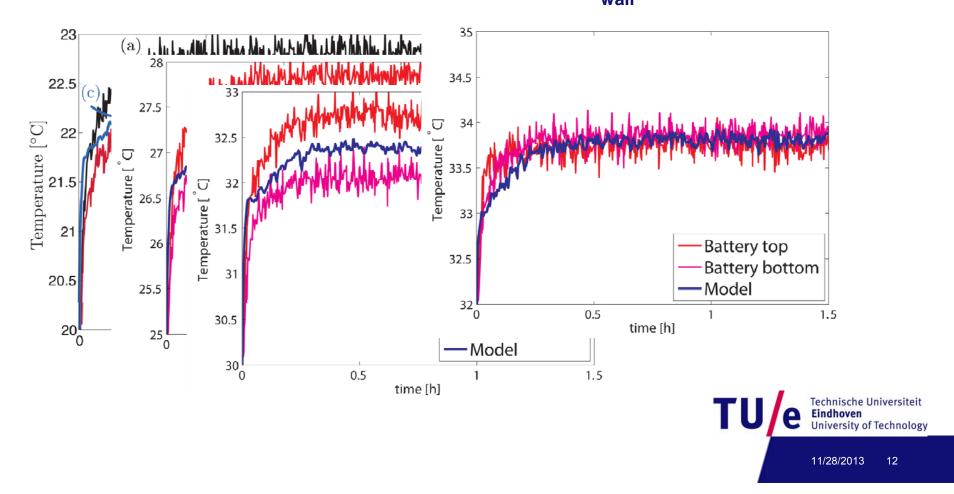
•  $T_{wall} = 30 \ ^{\circ}C$ 





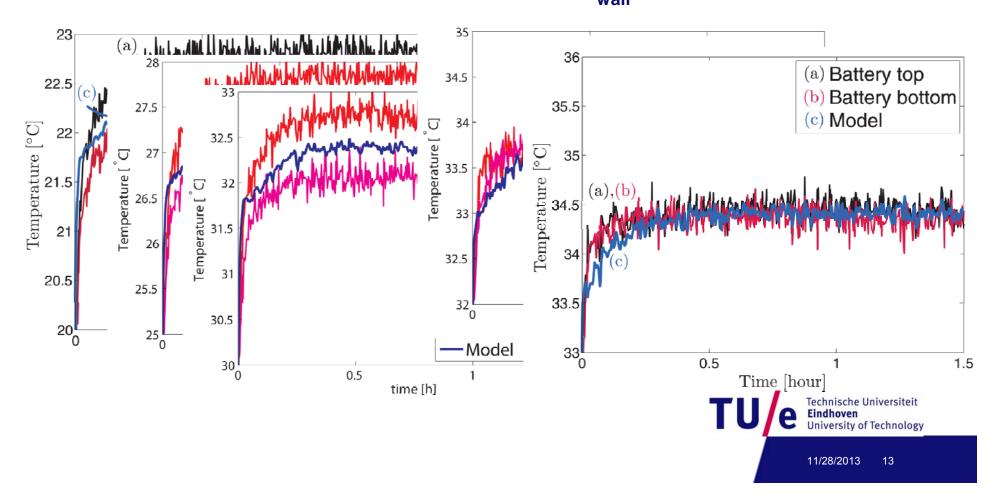
- Non boiling regime
  - $T_{wall} = 30 \ ^{\circ}C$

Boiling regime
 T<sub>wall</sub> = 32 °C

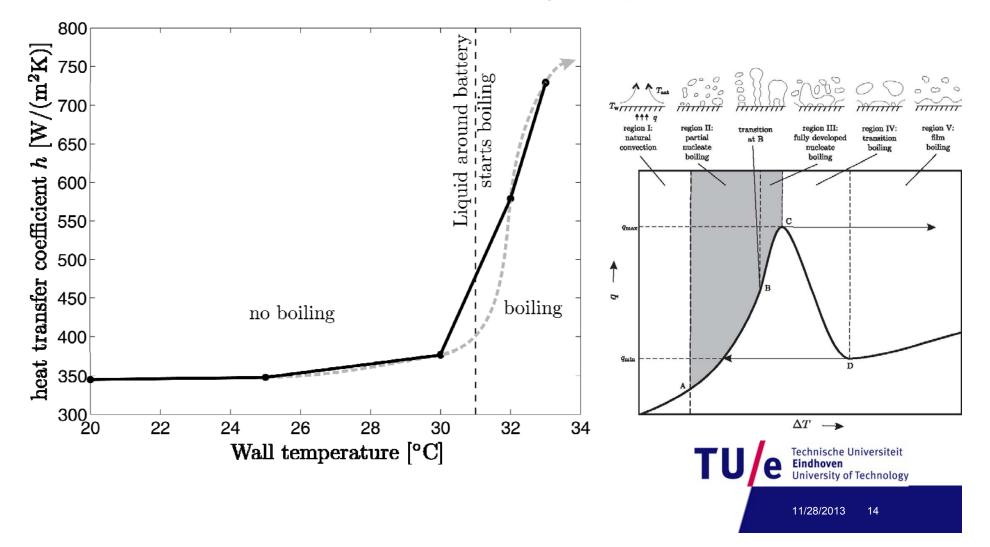


- Non boiling regime
  - $T_{wall} = 30 \ ^{\circ}C$

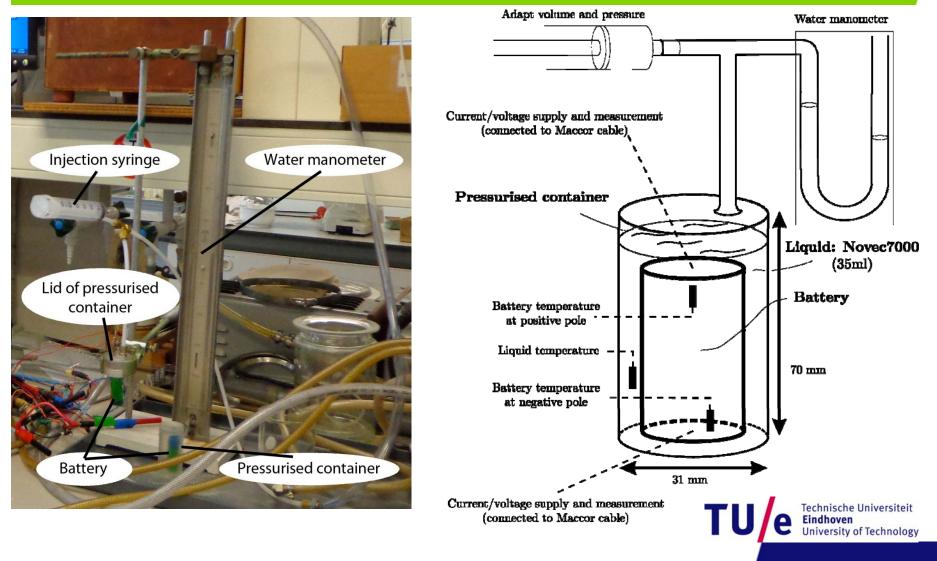
• T<sub>wall</sub> = 33 °C



Heat transfer coefficient from battery to liquid



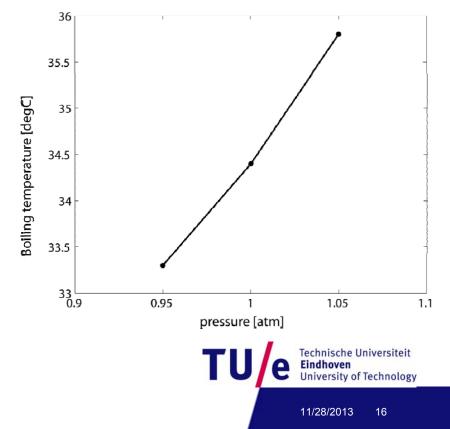
# Setup2



11/28/2013 15

#### **Experiments:** Boiling process control via pressure

- Requirement for application in Evs:
  - Actively and rapidly control the boiling process via the pressure:
  - Boiling temperature changes
    with pressure
  - Best results obtained directly after pressure change
  - Then the effect wears off due to the new thermal equilibrium the liquid and battery take



**Experiments:** Boiling process control via pressure

# 1 Ah Li-ion cell submerged in Novec7000 (3M) (BP @ 1 atm: 34C) PCD at 4.5A



# Conclusions

- This study investigates the ability of thermal conditioning by boiling
- Principal results are:
  - The working fluid (Novec7000) can be applied directly on the battery due to dielectric property
  - · Its cooling capacity greatly exceeds that of air
  - When boiling, the liquid can thermally homogenise the battery
  - Pressure can regulate the boiling process
    - Change pressure, instanteneously affects the boiling process
- Experimental proof of principle for thermal homogenization of battery packs in Evs
- Follow-up experiments are required

