

# KIVI / DOT Space for Subsea Saturation Diving, the Final Space Frontier?

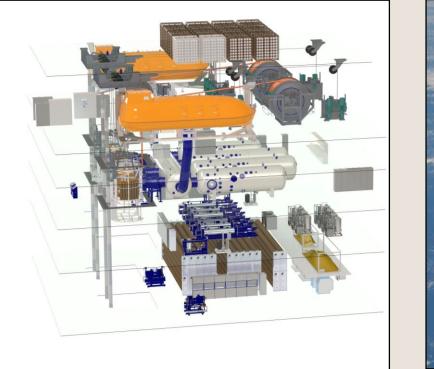
Johan de Bie, Royal IHC

IHC Hytech – KIVI-DOT Subsea-Space 2016 V 3-0

The technology innovator.



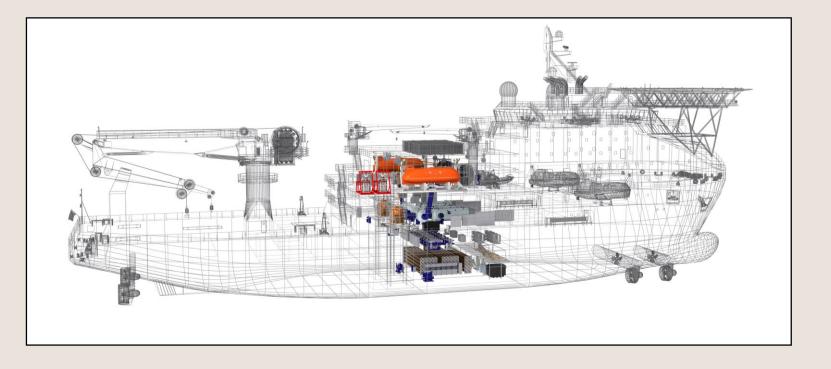
#### A place to live ; Comparison







#### A place to live ; Part of a larger structure



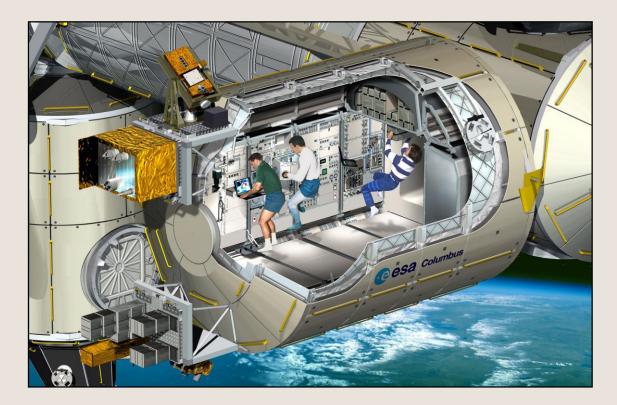


#### A place to live ; Saturation system



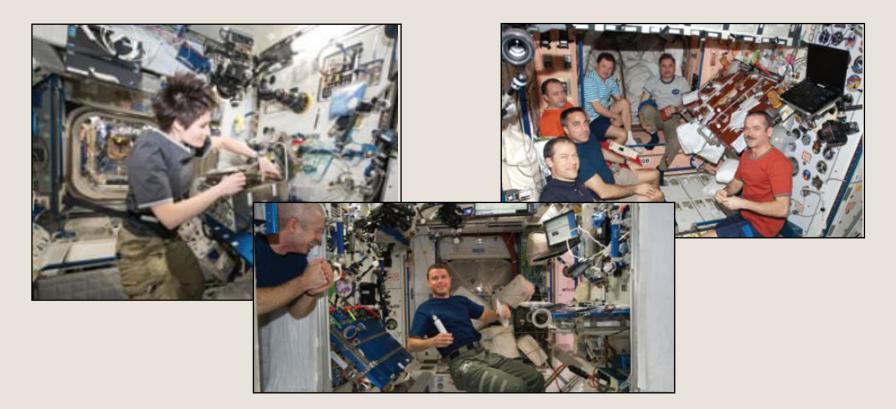


#### A place to live and work ; ISS



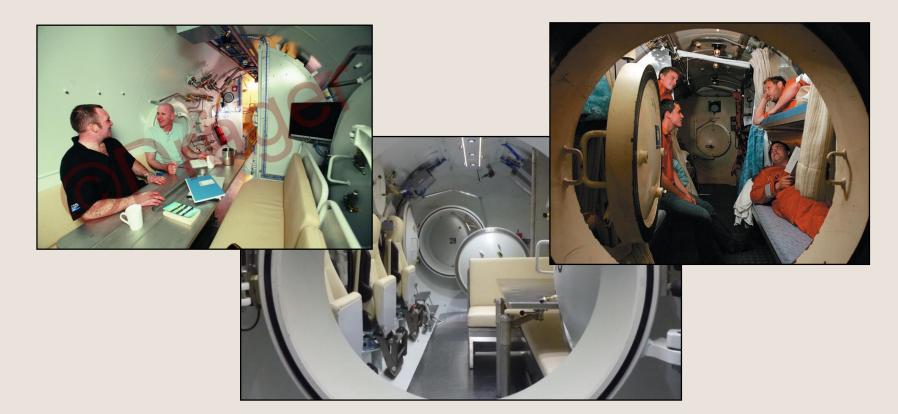


#### Inside the complex ; ISS





#### Inside the complex ; Saturation system





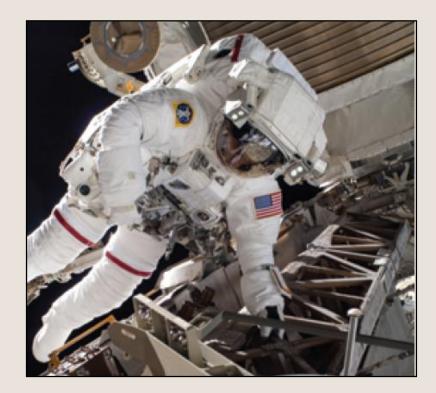
#### At work ; In the ISS





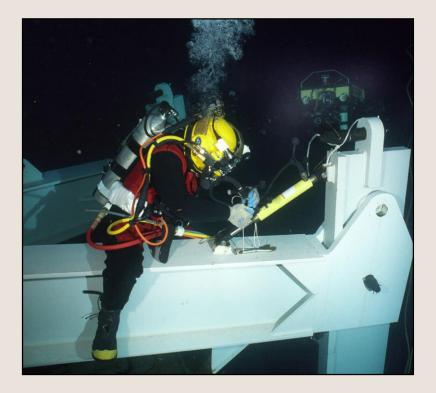
#### At work ; In the hostile environment







### At work ; Equipment similarities







#### Getting to work ; Saturation diving bell

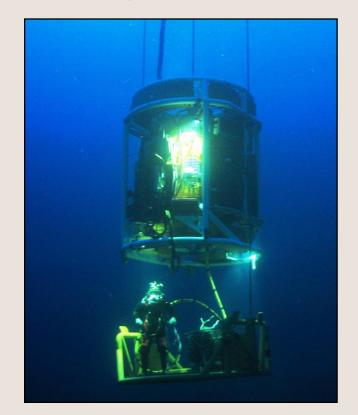






#### Locking out ; Saturation diving bell





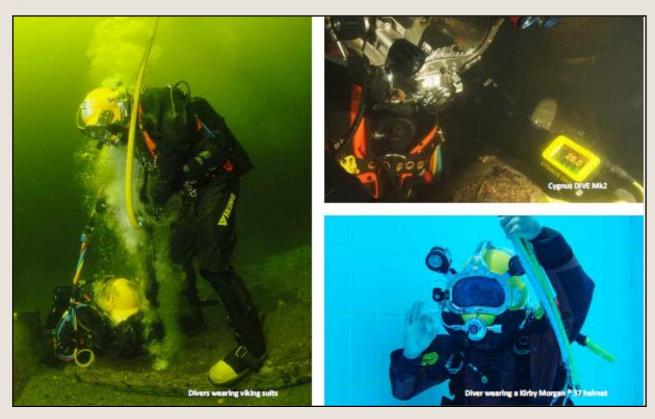


## Locking out ; ISS





#### Zero gravity ; Differences





#### Zero gravity ; Differences





# Zero gravity ; Different cases





## **Going home ; Astronauts**



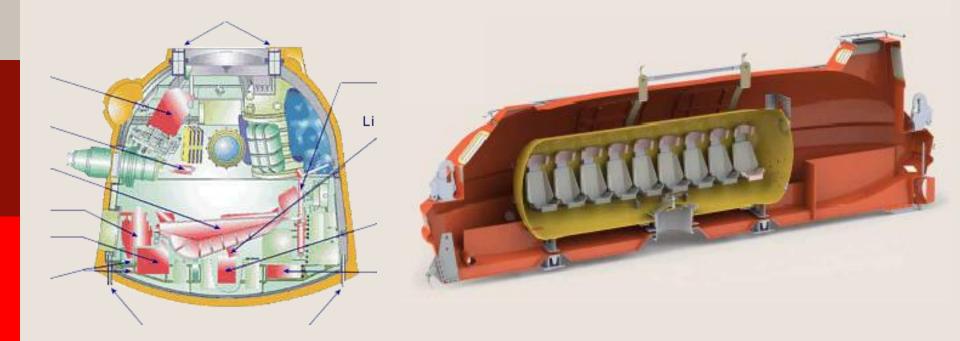


#### **Going home ; Saturation divers**





#### **Emergency escape ; Concept**





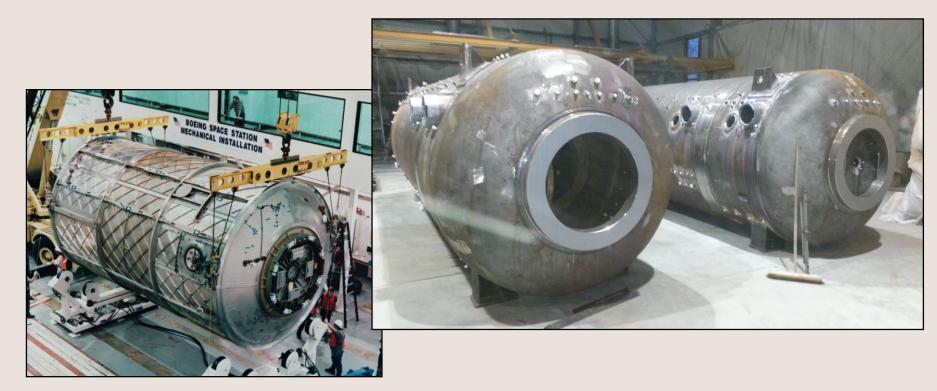
#### **Emergency escape ; After use**







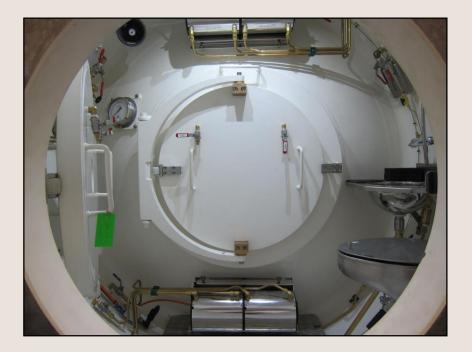
#### **Construction ; Pressure vessel**





#### **Construction ; Hatches retaining pressure**







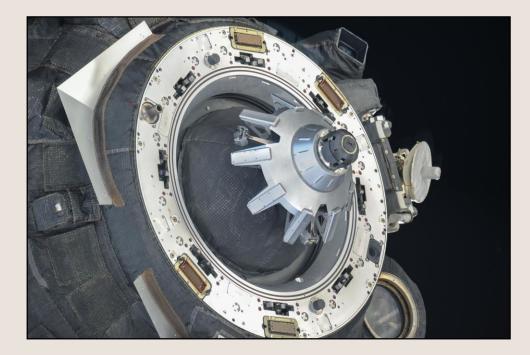
#### **Construction ; Viewports**







### **Construction ; Docking & Clamping**







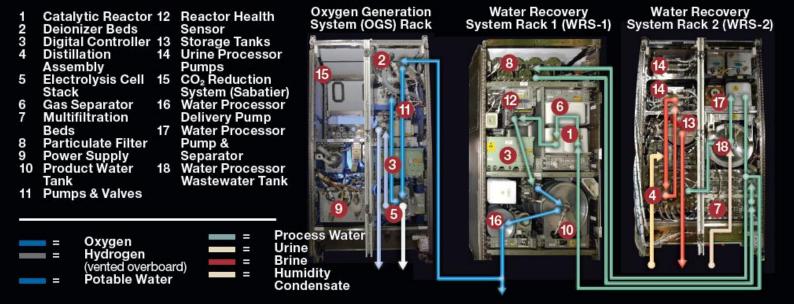








#### U.S. Regenerative Environmental Control and Life Support System (ECLSS)









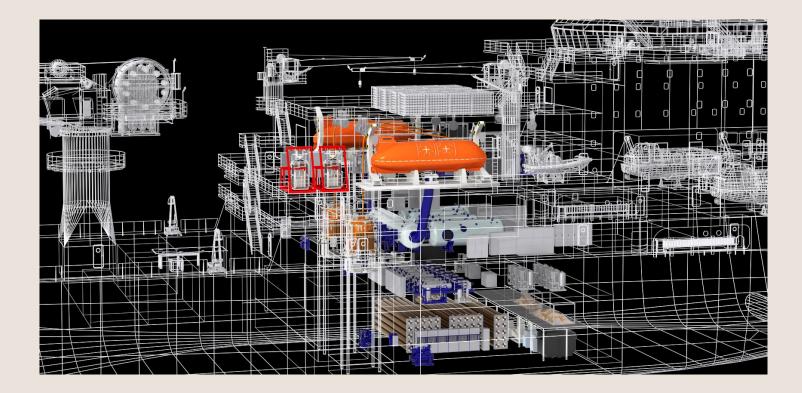








#### **Construction ; Supported from vessel**





#### **Construction ; Supported from earth**





#### **Construction ; Control rooms**





#### **Construction ; Control rooms**





#### **Breathable (livable) atmosphere**

#### Dangers surrounding a typical breathable atmosphere:

- Fire risk increases with higher Oxygen content
- **Pressure:** Pressure differences cause decompression
- **Toxicity:** Higher (pp) Oxygen content becomes toxic
- **Narcotic:** Higher (pp) Nitrogen content becomes narcotic
- **CO<sub>2</sub>:** Exhaled  $CO_2$  build up in enclosed environments

**Contaminants:** Contaminants build up in enclosed environments



#### **Breathable atmosphere**

#### Dangers surrounding a typical breathable atmosphere:

- **Moisture:** Moisture is a danger when living in enclosed environments
- **Bacteria:** Bacteria and germs can grow more easy in enclosed env.
- **Particles:** Debris, rubble, spills etc. can create dangers in encl. env.
- **Temperature:** Loss of body heat via breathing
- **WoB:** Work of Breathing more dense gas causes fatigue



**Space station:** 

#### Normal air at atmospheric pressure

Standard: 21% Oxygen (balance Nitrogen) @ 1 bar absolute

#### **Saturation diving:**

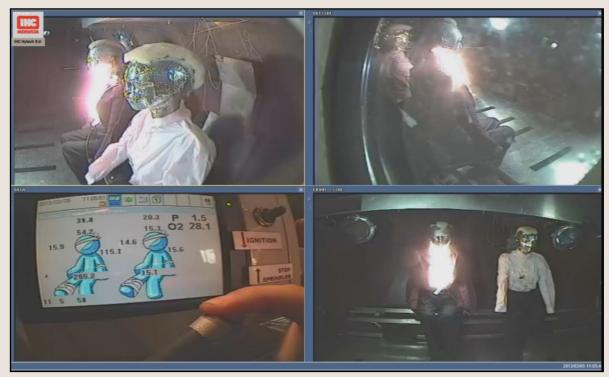
Heliox mixture with 0,4 ppO2 at required depth

Example: 2% Oxygen (balance helium) @ 190 msw (20 bar absolute)



Fire risk increases massively with higher Oxygen content

Also increases with enclosed spaces which can't be evacuated fully or of which life is dependent





#### Suppressing fire by exthinguising with:

Water (foam / spray / mist)

gas (CO<sub>2</sub> / other)

or other means







Contamination of breathing gas caused by fire





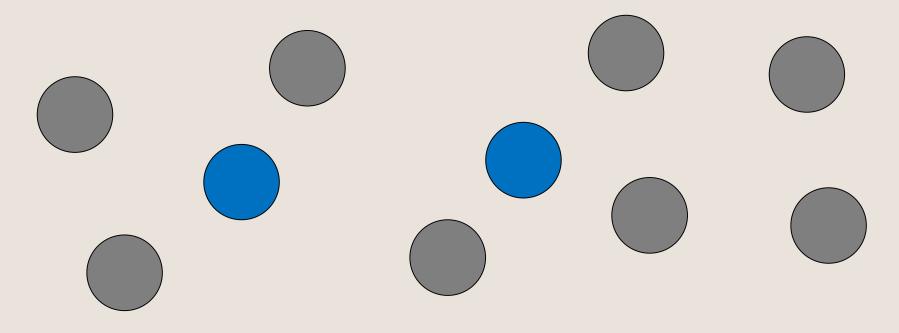
# NASA conducting fire tests live in space

Saffire tests onboard the Cygnus craft



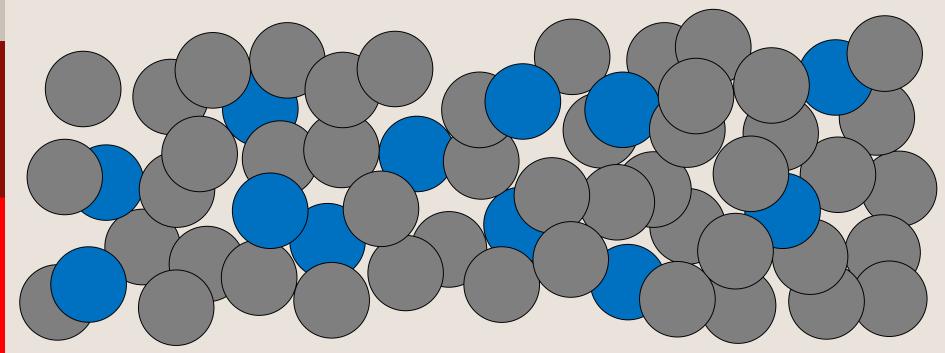


#### Normal air at atmospheric pressure



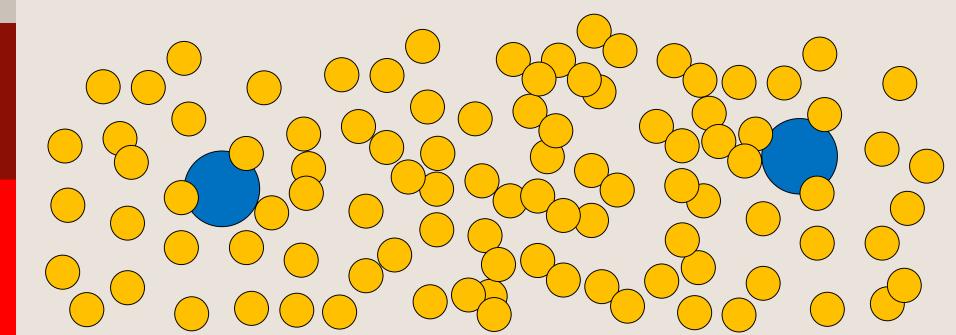


#### Normal air at 50 msw (6 bar absolute)



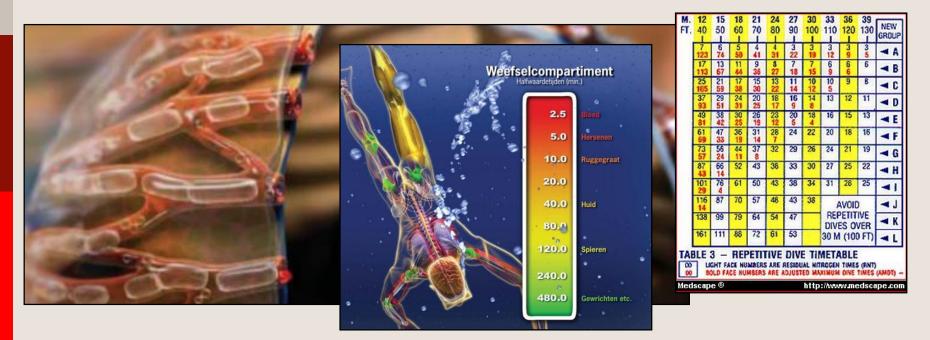


#### Heliox with 2% Oxygen at 190 msw (20 bar absolute)





#### **Decompression (DSI) when exposed to different pressures**





#### **Oxygen toxicity**

p <b>O</b> <sub>2</sub> (bar)	Time to cause 10% lung damage	Time to cause 20% lung damage
2.0	9 hours	15 hours
1.5	13 hours	20 hours
1.0	23 hours	Several days
0.8	Several days	
0.6	No damage	

#### Nitrogen narcotic starts at pN<sub>2</sub> of 3,2 bar (as of 30 msw)



CO<sub>2</sub> production by the human body:

Under normal working conditions metabolic usage of a human is 0,5 ltr/min of pure Oxygen

**Results in production of 0,5 ltr/min of pure CO<sub>2</sub>** 

Does not change under pressure !

Must be removed from atmosphere, toxic



CO<sub>2</sub> removal:

By scrubbing with sodalime, lithium hydroxide or molecular sieves

Alternatives being developed

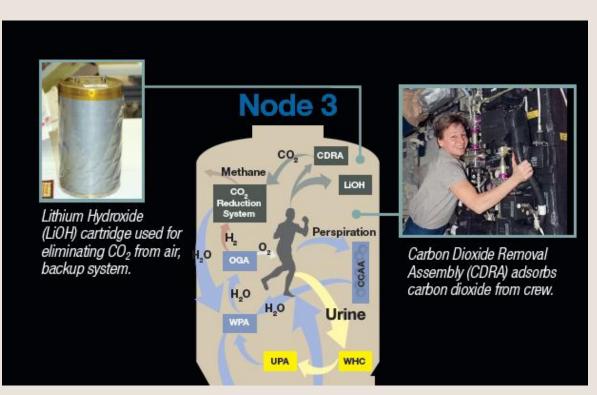




CO<sub>2</sub> removal:

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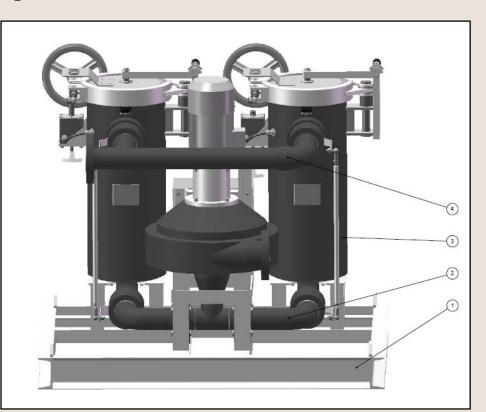




CO<sub>2</sub> removal:

By scrubbing with sodalime, lithium hydroxide or molecular sieves

Alternatives being developed





#### **De-Contamination:**

By scrubbing with blended carbon molecular sieves and catalists

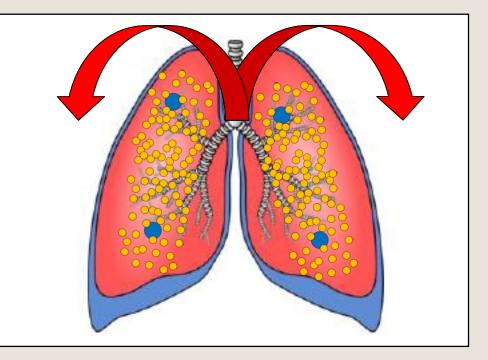
Removes pollution due to breathing, vomit, urine, feces, sweat etcetera





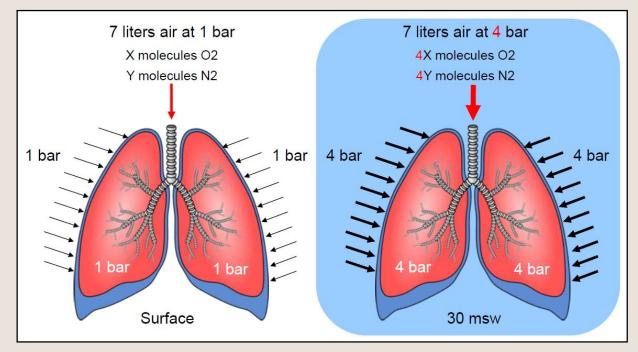
Loss of body heat via breathing, helium accelerates

Higher local temperatures inside a saturation dive system in order to prevent hypothermia





#### Work of Breathing, much more breathing gas handled





Lost particles present a hazard:

Contamination

Malfunction

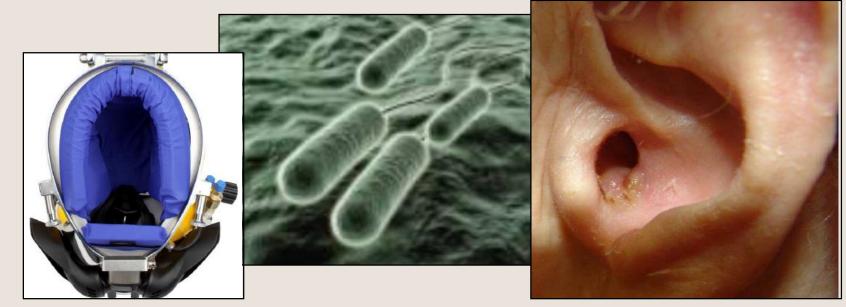
**Obstruction** 

**Short circuit** 





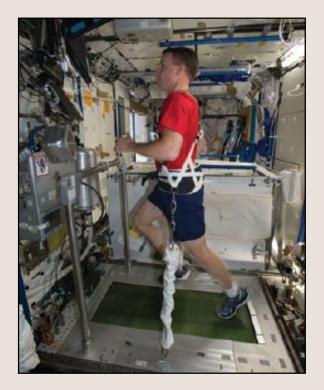
## Bacteria and germs can grow more easy in enclosed environment





## **Keeping fit**







## **Recovery of valuable supplies**

What is important to recover/remove for both systems, some examples given below:

Breathing gas; oxygen – helium – nitrogen

Fluids; moisture – toilet usage – condensation

Unwanted matter; bacteria – dirt – food spills

Environment; heat – cold – contamination



## Subsea and Space ; Best of both worlds !





## Information and images

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The technology innovator.